

A decorative network diagram in the top right corner, consisting of various sized circles (nodes) connected by thin lines (edges). Some nodes are solid grey, while others are hollow white with a grey outline. The connections form a complex, branching structure.

Lesson 5

Food Science and technologies

A decorative network diagram in the bottom left corner, similar to the one in the top right, featuring a mix of solid grey and hollow white nodes connected by thin lines.

Basic Components of Food

- ⊙ Water
- ⊙ Fats
- ⊙ Proteins
- ⊙ Carbohydrates
- ⊙ Acids and alkalis



Water

- ◎ Is the primary substance
- ◎ 95 percent of fruits and vegetables
- ◎ 65–80 percent of meats, poultry, fish
- ◎ Powerful solvent for many flavor compounds, vitamins and minerals
- ◎ Freezes at 32°F (0°C) and boils at 212°F (100°C)

Fats

- ◎ Some solid at room temperature, others, with less hydrogenation, liquid
- ◎ Dissolve certain flavor compounds, fat-soluble vitamins, and colors that water cannot dissolve
- ◎ Tenderize foods
- ◎ Fat smokes at 375°F (191°C) while oils smoke at 450°F (232°C).



oil begins to smoke when it gets very hot.

Proteins

- ⊙ Determine the texture of food
- ⊙ Proteins shaped like coils that react to certain items
- ⊙ Heat, salt, and acids make coils unwind, producing a soft texture and loose bonds
- ⊙ Tight bonds result in coagulation, forming a denser texture (like custard)
- ⊙ Kneading proteins produces tight bonds (bread dough)
- ⊙ Coagulation causes proteins to lose water



Figure 4-7 The Maillard reaction browns meats.

Carbohydrates

- ◎ Breads, pasta, grains, starchy vegetables, fruits
- ◎ Starch molecules soften in moisture
- ◎ Absorb moisture and swell, causing liquids to thicken
- ◎ Starch cells stick to one another and trap moisture
- ◎ Sugar attracts moisture by trapping
- ◎ Jams, jellies, fruit butters, and dried fruits are good examples

Acids/Alkalis

- ◎ pH range measures level of acid/alkali in food
- ◎ pH scale is 1–14, 1 being the most acidic
- ◎ 7 is neutral (water)
- ◎ 1–6 acid, 8–14 alkaline
- ◎ Foods being close to either end are not usually considered potentially hazardous

Heat Transfer


- ◎ Conduction, convection, radiation
- ◎ Conduction is the direct transfer of heat, pan on a burner or in an oven
- ◎ Convection is the transfer through gases or liquids (hot air, a fat or liquid)
- ◎ Radiation is the transfer of energy through electromagnetic energy
- ◎ Infrared radiation heats the surface of the food
- ◎ Microwave radiation uses high-frequency waves
- ◎ Induction energy uses coils that generate magnetic currents

How Cooking Affects Foods

- ◎ Changes colors
- ◎ Denatures proteins
- ◎ Improves flavor and texture if applied correctly
- ◎ Releases or destroys nutrients
- ◎ Caramelizes sugars, producing rich brown colors and flavors



The Maillard Reaction


- ◎ Protein food develops a brown color from heat application when proteins denature
 - ◎ Occurs in dry-heat applications
 - ◎ Caused by the amino acids in the food
 - ◎ Produces a wonderful flavor (if not overdone)
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Gelatinization

- ◎ Starch granules absorb water as they soften and cook
- ◎ Hold liquid in place
- ◎ Molecules join together
- ◎ Helps food hold shape
- ◎ Occurs at different temps for different types of starch
- ◎ Root-based starch breaks down faster in the presence of heat, sugar, and acids



Forming Emulsions

- ◎ Mixture of two substances not usually soluble
 - ◎ Object is to suspend one ingredient in the other
 - ◎ A vinaigrette is a perfect example of water/oil emulsion
 - ◎ Shake to disburse vinegar to oil
 - ◎ Butter is a water/oil emulsion
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Permanent/Temporary Emulsion

- ◎ Mayonnaise is an example of a permanent emulsion
- ◎ Egg yolks stabilize the emulsion
- ◎ Other stable examples are hollandaise and forcemeats
- ◎ Temporary emulsion example is oil and vinegar dressing, must be mixed constantly
- ◎ Other common emulsifiers are mustard, glace de viande, butter (fat)